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# AVIATION

## AND AERONAUTICAL ENGINEERING



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VOLUME VIII

Number 8

### SPECIAL FEATURES

THE AVORIO-PRASSONE KITE BALLOON  
STATIC PRESSURE GRADIENTS IN WIND TUNNEL WORK  
PRODUCTION AND MAINTENANCE OF AIRCRAFT  
THE CLAUDEL CARBURETOR  
THE LIFT OF HYDROGEN

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BY  
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# AVIATION AND AERONAUTICAL ENGINEERING

VOL. VIII, NO. 5

Member of the Audit Bureau of Circulations

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## THOMAS-MORSE AIRCRAFT CORPORATION



# The Avorio-Prasone Kite Balloon

By John Jay Ide

(Continued from p. 31, U. S. N. R. F.)

The Avorio-Prasone kite balloon was designed jointly by Major Luigi Avorio, Chief of the Aeronautical Division of the Italian Army, and Dr. Eugenio Prasone, Director of the Stabilimento Costruzioni Aeronautiche. It was used extensively for observation purposes in the Italian Army. Various and other valves were provided with A. P. balloons as a defense against Austrian aircraft. A number of these balloons were also used in England and used as a barrage around London.

In August, 1928, the U. S. Navy bought two A. P. balloons over to America. One was demonstrated at the naval air station, Rockaway, while the other was given to the Army and tested at the balloon school, Fort Omaha. Three similar balloons are being built by the Commercial Aircraft Corp. for the Navy.

The gas envelope of the A. P. balloon is disposed of around (depending upon the model) with the major and minor axes differing only slightly in length. To the rear end of the gas envelope there is attached an air-filled cone serving to reduce the wind resistance of the balloon and, in conjunction with the balloon, to maintain the shape of the gas envelope. Around this cone there are arranged three air flaps 122 deg. apart.

The wires and basket suspensions are attached to the same points on the envelope, which is the case of the model having the cylindrical gas envelope has the effect of bringing the basket and cone into very close together. The effect of this is to make the suspension of the basket to swing when the balloon is blown in very high winds. In the other model the gas chamber has been extended farther into the cone, increasing the air filling force at the stern thereby supporting the basket from the cable. By doing this a parachute basket can be used. If it is desired to bring the basket near the cable

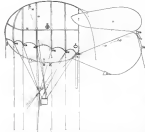


FIG. 2. DIAGRAM OF LONGITUDINAL CROSS-SECTION (1) WIND SHOCK TO BALLOON; (2) WIND SHOCK TO EXTEND CONE; (3) AIR VALVE ON CONE; (4) AIR INLET; (5) BURNING TO FIRE; (6) BURNING INFLATION; (7) AIR INLET; (8) BURNING INFLATION; (9) AIR INLET; (10) BURNING INFLATION; (11) BURNING INFLATION; (12) BURNING INFLATION; (13) BURNING INFLATION; (14) BURNING INFLATION; (15) BURNING INFLATION; (16) BURNING INFLATION; (17) BURNING INFLATION.

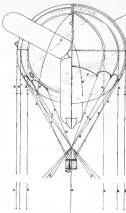


FIG. 3. TRANSVERSE CROSS-SECTION (LEFT) SIDE VIEW, (RIGHT) SIDE VIEW.

in the cylindrical model, it is merely necessary to change the lengths of the suspension ropes.

## Envelope

The gas envelope (Fig. 1) is composed of gases (as in the case of aeroplanes) and the spherical form is maintained by side and strong transverse bands (10) which at the vertices of the envelope, thus reinforcing the shaping points.

The basket main line is directly above the sternmost and provided in a line connecting the terminal rings of the ropes. Therefore, all the outer fabric on the outer side of the balloon, the part that rests on the ground when hoisted down, remains only air. The tension due to the lift and to the action of the wind, on the other hand, is all concentrated in the terminal rings and is transmitted to the cable through the suspension.

The sternmost are of steel wire cable 5 mm. in diameter, with a tensile strength of about 2000 lb. while the suspension ropes, made of hemp, are 12 mm. in diameter, with a tensile strength of about 2000 lb.

Three-ply cotton fabric, rubberized between the plies, is used for the gas envelope.

At the bottom of the gas chamber there is a diaphragm forming the balloon when the balloon is not longer full of



FIG. 4. AUTOMATIC GAS VALVE GEAR.

gas (in descent). Air is admitted into the balloon by means of the pump (1), the balloon also communicates with the bottom by the opening (4). An automatic gas valve (10), placed near the sternmost, is operated by means of small non-striking metal bars attached to the fibers of the diaphragm.

On top of the envelope is the opening (14). Opening this uncovers a series of elliptical holes for the escape of the gas. The opening and cone through the gas chamber, out the port side of the envelope and thence to the rear.

## Cone and Flap

The cone and flap are filled with air which enters the lower flaps through the pump (1) and the opening (4). From the balloon (Fig. 1) the opening (14) between the lower flaps and the cone are filled with fabric valves allowing the air to pass from the flaps to the cone but not back again. Thus after the cone and upper flap is inflated by a pump through the air inlet valve (1), thus allowing escape at the beginning of

the ascent, while the lower flaps remain deflated until wind opens by the wind during the ascent.

The fabric valve (3) at top of the cone is so adjusted that it begins to open at a lower pressure than the gas valve. The suspension of the gas valve opens against the cone, air from the cone and then at the same time as from the cone.

On starting, complete air deflation is obtained in a very few minutes by opening the proper valve.

Two-ply cottonized cloth fabric of very fine texture is used for the cone and flap.

## Suspension

The cable suspension (Fig. 2) consists of twelve ropes which start from the vertices of the sternmost on the sternmost and come together in two groups, right and left, each group terminating with a metal ring. The two rings are joined together by means of a ring on a steel rope to which the main cable is attached.

Two further rings are also attached to these mentioned above. If the basket is near the cable one of these rings carries the ropes for the control and bag attachment and also the control handling ropes, while the other takes the two forward suspension ropes of the cone. Otherwise, the steel bag



FIG. 5. INFLATION OF AN A. P. BALLOON.

attachment and handling ropes are attached to the two rings, the forward rear suspension ropes running to the sternmost.

The cone suspension consists of steel ropes of which the rear side and the two sides are attached to the terminal rings at the sternmost on the envelope. If the basket is near the cable the two rear ropes are secured to the sternmost rings. If not, the two rear ropes are attached to the sternmost rings. The cone suspension ropes are marked with colored thread and are joined to the cone ropes having the same color and, with numbers. All suspension ropes, cable and rear side, are tied with clamps so that their lengths may be adjusted.

## Gas Valve Adjustment

During the inflation of the balloon the closed sternmost device, about 60 in. long, is kept down and, when it begins to fill with gas, it is stopped. It is then possible to see when the balloon is full and the adjustment of the valve (Fig. 3) can be provided with, setting the hook of the last of the cone cone so that the valve will open at the proper pressure. As the hose ends are of steel, once the adjustment is made the valve will not shift, sufficiently to affect the operation of the valve.

## Gas Valve Construction

A balloon, when set up slowly, is at least potential as a field of very different potentialities, thus constituting a large potential difference between the balloon and the surrounding air. Under these conditions certain "normal" discharges occur from the surface due to the movement of the balloon, especially from projecting points. The flow of electricity from the balloon to the surrounding atmosphere gives rise to discharges, which although slight, are capable of producing the ignition of a mixture of hydrogen and air.



FIG. 6. GAS VALVE OF AN A. P. KITE BALLOON.











## The Claudel Carburetor

The Claudel carburetor, widely used in Europe for twenty years, is now made in America by the Claudel Carburetor Co., Long Island City, N. Y. This highly-protected, plane-table mixing device was designed by Charles Henry Claudel, the pioneer in the development of the plane-table type of carburetor, who is recognized as the foremost European authority on carburetors.

Recently made by the Claudel carburetor in European design, before the war were both numerous and comprehensive, twenty-three types having been captured in 1923, in addition to the Italian-made one, and the breaking of all world's records on the Brooklands track in England. As a result, Claudel carburetors were employed exclusively on the famous Allied airplanes engines from the beginning of the war in 1914 until the end. Among these were the Delle-Reyes, Redoubt, Peugeot, Hispano, Hispano-Suiza and Renault.

Claudel carburetors made the first record trip across the Atlantic from England to America on the Hispano engine.

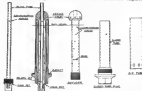


FIG. 1. DIFFUSER OF CLAUDEL CARBURETOR IN DIFFERENT POSITIONS.

of the British de Havilland G.34. They were also used on the Koda Koyan engine of the Vickers-Vimy airplane which was the first plane to make a nonstop flight across the Atlantic. The fact that this craft completed its long flight with one-third of its fuel secured in a striking commentary on the fuel-saving ability of the Claudel.

Another record held by the Claudel carburetor is that for power and speed established by Rich Leveque, the famous French aviator, who set a new world's speed record of 232 miles per hour.

### Automated in Design

The engineers of the Claudel Carburetor Co. have incorporated the European model of the Claudel to meet the particular requirements of engineering needs in this country. They have added several features demanded by the American market, such as a spark starting device and rapid acceleration with a cold engine.

### The Claudel Carburetor

Early carbureting devices employed a spring-controlled air valve in an effort to secure the proper mixture between throughout a wide range of engine speeds. In 1903 Charles Henry Claudel, of Paris, France, patented the first table automatic carbureting carburetor without the use of moving parts. His early principle of breaking up the gasoline by swift current of air, making an emulsion inside the jet itself before delivery to the carburetor proper was original with him and has since been widely copied. The modern Claudel retains the same principle, refined and improved to reproduce the heavy fuel of today.

### The Claudel Principle

The automatic function of the Claudel carburetor is based upon the action of the diffuser jet assembly. This device combines three distinct principles of operation:

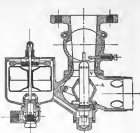


FIG. 2. SECTION THROUGH THE STAYFAST AMERICAN DIFFUSER BY CLAUDEL CARBURETOR.

First.—The orifice or jet which supplies all of the liquid fuel is placed at charge, or below the fuel level, when at low to proportional to the diffuser in level.

Second.—The diffuser proper is in an "ideal" position, that is, enclosed in a chamber with openings at one end to the atmosphere and at the other end to the surface on the carburetor, thereby reducing the effect of the suction on the fuel flow.

Third.—Through the nozzle hole in the diffuser head the interior venturi chamber of the diffuser is placed under the



FIG. 3. GENERAL VIEW OF THE AMERICAN CLAUDEL CARBURETOR.

lowest section of the vacuum in the carburetor, maintaining the discharge in the vacuum is increased.

The combination of these three principles of operation, correctly applied, gives a balanced mixture proportioned for the best results at every engine speed and load.

The Claudel is really a combination of two types of jet, one, the main jet, Fig. 3, of constant force per unit of fuel, the other, shown as a series of nozzles below, of variable-extended flow. Fig. 1 indicates the fuel level when the carburetor is at rest. Fuel from the fuel chamber flows through the main jet into the diffuser, then through the overpassing holes into the jet orifice or diffuser hole. It also flows through the sliding jet into the sliding hole, the level in the three tubes being the same in that in the first chamber.

### The Diffuser

As at atmospheric pressure enters the outside hole or air hole of the diffuser orifice, passes up the main duct and over the top of the gasoline guard tube, which prevents the



FIG. 4-6. DIAGRAMS OF MIXTURE FLOW OF CLAUDEL CARBURETOR. LEFT: DIFFUSER THROUGH IDEAL, IDEAL JET IN POSITION. MIDDLE: AIR FLOW THROUGH, DIFFUSER JET IN POSITION. RIGHT: THROUGH WIDE OPEN.

fuel from overflowing. As the throttle is opened gradually and the mixture in the diffuser increases, thereby lowering the liquid level in the diffuser hole, a series of air bleed or overpassing holes are progressively uncovered. Through these holes the air passes into the ascending column of gas vapor and out the main hole at the top of the diffuser in a fairly under-up gasoline emulsion. The higher the mixture rises upon the diffuser, the lower will be the level of gasoline within it, and therefore, more of the overpassing holes will be uncovered, producing a greater dilution of the mixture.

At the higher levels the diffuser is gradually supplied and twenty-one rapidly-arranged air bleed holes are in action. As the gasoline gradually rises filled by the mixture, they enter from through a series of twenty-one air bleed holes in the right angle into the ascending fuel. This venturi effect produces a fairly divided fuel emulsion ideal for speed, power, economy, and high velocity. Also, the diffuser discharge is in approximately the same of equal parts of air to one part of gasoline by volume. This through combination of air and gasoline before it enters the carburetor barrel in the Claudel method of breaking up the heavy fuel of today for mixture economy.

It will be easily seen that any kind of a power or consumption error desired can be secured by changing the size and position of the overpassing holes in the diffuser wall. For example, if the consumption error shows a rich spot at an engine speed of 1000 r.p.m., the holes at this level of the gasoline in the diffuser could be enlarged, throwing out the mixture and bringing it to the point desired. In this way, an exact gasoline and air ratio can be maintained for the best results.

The clean, straight, vertical of the Claudel carburetor, with the barrel throttle giving an extended vacuum effect, leaves no obstruction in the path of the charge and consequently full volumetric efficiency is achieved. Also, the gasoline layer of the interior walls inside the mixture is uniform in composition and velocity.

### The Sliding Device

The sliding device is incorporated in a central tube projecting up into the depression of the barrel throttle when a strong pull is exerted on the sliding jet for low-speed action. The barrel throttle is closed in position around the sliding jet and the only adjustment on the carburetor is to serve, entering into the air space to partially block off the area of this jet as desired, decreasing it as in excess the air area and operates the sliding mixture. Increasing it and makes the mixture leaner.

### Easy Starting Feature

In the American type of Claudel carburetor shown in Fig. 5, a decided improvement has been incorporated for starting the jet by means of a sliding air cone, mounted inside the duct. This device secures quick starting and warming up in cold weather, and absolutely discourages idleness. The automatic air cone, combined with the diffuser, may be raised into contact with the jet, making off the air supply and putting the fuel section on the diffuser. In this position

it is only necessary for the cylinders to exhaust the air in the top of the carburetor before the diffuser discharge next commences. This action is instantaneous, as the vacuum is small.

Referring again to Fig. 2, the sliding cone A, in its closed or stopping position, fits snugly into the varying jet of the carburetor. The shape of the cone makes the air stream so that it passes the nozzle hole of the diffuser with high velocity and in an even column ideal for the best results. In its lowest position the cone offers no restriction to the maximum performance required and so of benefit or consumption in idle conditions.

In warming up a cold engine when a richer mixture is necessary, the cone A may be dropped slightly, leaving a small hole open. With the cone in this restricting position some may readily get away while the engine is still cold. Thus, in reality, the Claudel acts as a double carburetor—one of small volume for normal driving, and, with lowered cone, number of full engine capacity.

The advantage of the Claudel cone over the common type of battery strangle valve, placed some three or more inches ahead of the fuel discharge nozzle, is very evident. In the latter type, all of the dead air lying between the nozzle and the strangle valve must be exhausted before fuel will begin to flow. In the Claudel type, mixture on the nozzle is instantaneous.

When the common type battery strangle is only slightly open (usually being warming up the small amount of air passing the fuel discharge nozzle has little or no velocity and hence can push up very little fuel. Fuel, however, in discharge, the common type carburetor, is discharged in a chamber than in the mixing chamber, and instead of being picked up by the air, the raw fuel runs down to the battery strangle, where the air velocity is higher and is so partially entrained and carried into the engine. In the Claudel type, on the other hand, no emulsion is discharged



## The Lift of Hydrogen

By Squadron Leader P. L. Todd, R. A. F.

The *True Lift* as applied to airships is equivalent to buoyancy, and this buoyancy, when applied to water-borne ships, is known as *net buoyancy*.

(1) The weight of the ship.  
(2) The weight of supporting medium displaced by the ship.

Here the analogy between water and air borne craft ceases. For in the case of water-borne ships (other than the submarine) a condition of equilibrium always prevails, if the weight of the ship is increased its displacement is increased, the ship being lowered, but a state of equilibrium is maintained.

The analogy, being entirely surrendered by the supporting medium, but is equivalent in the atmosphere in water has produced equilibrium. If the change in weight of the displacing supporting medium the analogy may possess either positive or negative buoyancy.

While the structural weight of the ship remains constant, the weight of medium displaced by the ship is subject to considerable variation.

The following affect the buoyancy—

- (1) The volume of gas in the airship.
- (2) The purity of the gas.
- (3) The barometric pressure of the air.
- (4) The temperature of the air.
- (5) The amount of water in the air.

Of the five causes affecting the lift of an airship apart from its volume, the purity of the gas is generally to be taken as the most significant factor, but the combination of temperature and barometric pressure may produce even more important variations.

The maximum purity of hydrogen is an airship does not contain 99 per cent by volume, while certain chemical compounds accurately approximate the density of the airship at 99 percent by volume.

Under mean atmosphere conditions, air temperature 59 deg. Fahr. and barometer 30.1 in. Hg, the lift of 1,000 cu. ft. of hydrogen at 99 per cent purity is 69.7 lb., while under the same atmosphere conditions at 99 percent purity it is 68.6 lb.—a variation of 1.1 lb. or 1.6 per cent.

Now turning to purity atmosphere effects, 1,000 cu. ft. of 99 per cent pure hydrogen at a barometric pressure of 30.1 in. Hg and a temperature of 59 deg. Fahr. gives a lift of 69.7 lb. while an equal volume of gas at the same purity at a barometric pressure of 28 in. and a temperature of 90 deg. Fahr. gives a lift of 62.2 lb., a variation of 7.5 lb. per 1,000 cu. ft. of volume.

Herein illustrated the variations in lift due to (1), purity, and (2), to atmospheric changes, it is perhaps permissible to state that while the lift of 1,000 cu. ft. of hydrogen is 69.7 lb. at a temperature of 59 deg. Fahr. and under a barometric pressure of 30.1 in. Hg, the lift of 1,000 cu. ft. of hydrogen 99 per cent pure at a temperature of 90 deg. Fahr. and under a barometric pressure of 28 in. Hg, is a variation of 27.5 lb. per 1,000 cu. ft., which represents the amount of purity and atmospheric conditions which may be met with.

Having dealt with the general principles of lift and illustrated its variations by some examples, the question will now be dealt with in detail.

### Purity of the Gas

It is customary to give the chemical composition of the hydrogen as an example in as many per cent pure gas. This term means the percentage of hydrogen contained by volume, assuming the impurity to be of the same density as air.

Now it will be seen that if the impurity in the hydrogen is less than pure air in weight, the relative lift of equal volumes of impure hydrogen under the same atmospheric conditions is in the same ratio as their hydrogen content (by volume). It may be said: The lift of unit volume is directly proportional to purity.

### Aluminum in Lift by Barometric Change

One thousand cubic feet of pure hydrogen at 90 in. barometric pressure and a temperature of 40 deg. Fahr. weighs 3.55 lb., 1,000 cu. ft. of dry air under the same atmospheric conditions weighs 90 lb. Therefore, on the basis lift is the difference between displaced weight and displacing weight:  
90 — 3.55 lb. = Lift of 1,000 cu. ft. of pure hydrogen at 90 in. Hg. barometer and 40 deg. Fahr.  
= 86.45 lb.

Now assume that the temperature remains constant but that the barometric changes to 80 in. Then by the application of Boyle's Law it will be seen that:  
90 x 21 = 3.55 x 21 = Lift per 1,000 cu. ft. of 100 per cent pure hydrogen at 21 in. barometer and 40 deg. Fahr.  
= 82.8 — 3.72 = 79.08 lb.

But the above is an inaccurately completed way of arriving at the desired result, for the mathematical process may be simplified to —

|                 |           |
|-----------------|-----------|
| 31              | 30        |
| — (80 — 5.05) = | — (74.45) |
| 30              | 30        |
| =               | 79.05 lb. |

But 74.45 lb. is the lift of 1,000 cu. ft. of hydrogen 100 percent pure at 30 in. barometer and 40 deg. Fahr. Therefore the rule may be expressed: The lift of the unit volume is directly proportional to barometric pressure.

### Aluminum in Lift by Thermometric Change

By the direct application of Charles' Law it can be shown that: Lift of unit volume is inversely proportional to the absolute temperature.

To ascertain the three rules which have already been stated:

- (1) Lift of unit volume is directly proportional to purity.
- (2) Lift of unit volume is directly proportional to barometric pressure.
- (3) Lift of unit volume is inversely proportional to absolute temperature.

From the above the following formula may be deduced to give the lift of 1,000 cu. ft. of hydrogen under any conditions of purity, temperature or barometer.

74.45 x per cent purity x barometer x 1000  
= Lift in lbs.

On. Ft. in the —  
100 x 90 x (460 plus temperature in degrees Fahr.)

12.61 x per cent purity x barometer in inches

496 plus temperature in degrees Fahr.

The Temperature of the Gas

In the formula which has been given it has been assumed that the temperature of the gas is the same as the temperature of the air—i.e., a condition which is possible in a water-borne ship, but during flight there is frequently a material difference in temperature between the two. This difference arises in two ways—

- (1) The effect of sun on the gasbag.
- (2) The effect of a quick descent after flying for a considerable time at a high altitude.

In the first case the sunlight striking the gasbag causes some rise in temperature, the heat being in some of the form transferred to the gas itself. The gas then expands in accordance with Charles' Law. Now, in rapid descents if the pressure is not full, or in unsteady shape of the air in the balloons, this expansion will cause a greater volume of air to be displaced and consequently an increase in lift is obtained.

For example: At 40 deg. Fahr. and 100 percent purity

1,000 cu. ft. of hydrogen weighs 3.55 lb., and 1,000 cu. ft. of air under the same atmospheric conditions weighs 90 lb. Hence the lift per 1,000 cu. ft. is:

90 — 3.55 lb. = 86.45 lb.

Now assume that the temperature of the gas rises to 90 deg. Fahr. Its volume instead of being 1,000 cu. ft. will be:

1,000 x 520 = 1,040 cu. ft.

But the weight is still 3.55 lb., although the weight of air displaced is not the same; for instead of 1,000 cu. ft. being displaced, 1,040 cu. ft. are displaced, which weight:  
93 x 1,040 = 96.72 lb.

is 1,000 — 96.72 = 83.28 lb.

So the lift of the original 1,000 cu. ft. of gas becomes:

83.28 — 3.55 = 79.73 lb.

This alteration in lift by difference in temperature of the gas may at first appear to be of more importance in the case of large airships than in the case of small ones. Take the case of a Zeppelin of 2,500,000 cu. ft. maximum capacity containing only 1,000,000 cu. ft. of hydrogen. The gas, being in equilibrium when the sun comes out and raises the gas temperature to rise from 50 deg. Fahr. to 60 deg. Fahr. The lift of the airship is then 2,500,000 cu. ft. divided by 1.18, but the volume of gas was assumed to be 1,000,000 cu. ft. therefore the lift of the ship will be increased:

1,000,000 ÷ 1.18 = 847,457 cu. ft.

This alteration in lift during flight of large airships is of considerable importance, and it is consequently necessary that the pilot should always be acquainted with the relative temperature of the gas and air, for which purpose reliable thermometers are now fitted inside the gasbag, recording in the air.

The Amount of Water in the Air

Though it may not appear of importance, the humidity of the atmosphere has a certain effect on the lift of hydrogen. The greater the amount of water vapor in the atmosphere the less the weight of unit volume of air consequently the less the lift of unit volume of hydrogen, for the difference between displaced and displacing weight is less.

This alteration in lift due to humidity is a very seasonal, as it does not generally cause a difference in lift of more than 7 lb. per 1,000 cu. ft. of gas. However, in the case of "ball" balloons some adjustment may be made for humidity. This adjustment can most easily be made by observing the difference in temperature between the wet and dry thermometers and then consulting an empirically compiled table which gives the necessary adjustment for 1,000 cu. ft. of gas.

The Lift of Airships

So far the lift of hydrogen and the phenomenon which affect it only have been dealt with. Now the lift of airships will be considered. Before dealing with this two terms require definition: *net*, *available* lift, the others, *Displaceable Lift*, *Available Lift* is the total lift of the gas in the airship. *Displaceable Lift* is the difference between the available lift and the fixed weight of the airship. The *Displaceable Lift* is that buoyancy which is available for carrying fuel, oil, crew and ballast.

As will be seen, both the *Available Lift* and the *Displaceable Lift* will vary with temperature, humidity, humidity and purity. Consequently a gasbag can be regarded as standardizing on their weight 20 lb. per 1,000 cu. ft. of gas, the gas being 20.5 lb., the thermometer 50 deg. Fahr. and the air dry.

From the formula for lift already given, or by means of the *net* lift table, it can be calculated that the lift of 1,000 cu. ft. of gas under these conditions is 97.7 lb. per 1,000 cu. ft. Thus the tonnage of existing classes of English airships are—

60 x 97.7 tons = 5,862 tons.

S. S. class 2240

|             |            |      |               |
|-------------|------------|------|---------------|
| C. P. class | 188 x 62.7 | tons | = 8,59 tons.  |
|             | 2460       |      |               |
| N. S. class | 300 x 67.7 | tons | = 32,36 tons. |
|             | 2240       |      |               |

In building airships there *Displaceable Lift* under the conditions of purity, humidity, barometer and temperature already mentioned is generally specified but when the *Displaceable Lift* is taken positively, it is improbable that all, if any, of the specified conditions will occur consequently the *Displaceable Lift* must be corrected by simple mathematics to what it would have been had the conditions of the specification prevailed.

To convert *Displaceable Lift* from one set of conditions to another the following may be known—

- (1) *Displaceable Lift* on the basis of the total.
- (2) Temperature on the basis of the total.
- (3) Purity of the gas on the basis of the total.
- (4) Humidity on the basis of the total.
- (5) Humidity of the air on the basis of the total.
- (6) Volume of the gas or fixed weight of ship on the basis of the total.

To take an example: Let it be assumed that by practical trial it was found to be the *Displaceable Lift* of a 1,000,000 cu. ft. airship, 100 per cent full of hydrogen 99 per cent pure, when the barometer was 30 in., the temperature 40 deg. Fahr., and the humidity of the atmosphere negligible. It is required to know what would be the *Displaceable Lift* under the generally specified conditions, namely 100 per cent full of hydrogen, 99 percent pure, when the barometer is 20.5 in., the temperature 50 deg. Fahr., and the humidity negligible.

The first thing to be done is to determine the total lift as the basis of the practical trial.

By means of the lift formula already given, or by means of the *net* lift table, it can be calculated that the lift of 1,000 cu. ft. of gas on the basis of the trial. This will be found to be: 74.45 lb. per 1,000 cu. ft.

Now the volume of the ship is known to be 1,000,000 cu. ft. Therefore 74.45 tons is the *Available Lift*.

Therefore 2240 — 74.45 tons is the *Displaceable Lift* of the ship was found by trial to be 2 tons.

Therefore 32.50 — 74.45 tons is the weight of the ship. = 22.50 tons.

Now by means of the formula or table already mentioned the lift per 1,000 cu. ft. of hydrogen under the conditions of the specification. This will be found to be: 97.7 lb. = 49.85 tons.

Therefore under conditions of the specification 49.85 — 22.50 tons is the *Available Lift*.

Therefore 2240 — 49.85 tons is the *Displaceable Lift* under specified conditions. = 6.53 tons.

If the fixed weight is known but not the volume. As the example which was taken, the volume of the ship was known but not its fixed weight. Now let it be assumed that the weight was known but not the volume. Under these circumstances, if in other respects the same problem was to be solved, the procedure would be:

Determine the lift per 1,000 cu. ft. on the basis of the conditions of the trial. This would be: 74.45 lb. per 1,000 cu. ft.

Now the weight of the ship was known to be 22.50 tons. Therefore 22.50 plus 74.45 tons is the *Available Lift* of the ship. = 39.94 tons.

Therefore 22.50 x 2240 — 39.94 tons is the *Displaceable Lift* of the ship.

Therefore 74.45 — 39.94 tons is the *Displaceable Lift* of the ship.

= 1,000 thousand cu. ft.

= 1,000,000 cu. ft.

If neither volume nor fluid weight are known.

Operation may also be used when it is desired to know the Disposable Lift of an airplane when the conditions of the specifications do not state and require the volume and the weight of the ship is known.

Let it be assumed that on the occasion of the practical trial the Disposable Lift was found to be 2 tons when the ship was 100 per cent full of gas of 60 per cent purity and the barometer was 29.5 in. and the temperature 50 deg. Fahr.

Determine the lift of 1,000 cu. ft. of hydrogen under these circumstances by formula or table. It will be found to be 72.5 lb.

When this has been done, it is necessary to wait until some of the conditions have changed. Assume on a later occasion that the ship is 100 percent full with hydrogen 95 per cent pure, when the barometer is 29.5 in. and the temperature 60 deg. Fahr. Take the Disposable Lift, which for purposes of argument will be assumed to be 5 tons.

Then work out the lift per 1,000 cu. ft. of hydrogen under the conditions prevailing.

This will be found to be—

64.7 lb. per 1,000 cu. ft.

Between the first and the second trial the Disposable Lift has altered from 5 tons to 3 tons, and the lift per 1,000 cu. ft. from 72.5 to 64.7 lb.

Now the difference per 1,000 cu. ft. in lift as the two occasions is 72.5-64.7 lb. = 7.8 lb.

The difference in Disposable Lift is—5 tons.

= 4 tons = 8000 lb.

Therefore since the fluid weight of the ship has remained constant,

8000

= in the volume of the ship in thousands of cubic feet.

1148.7 thousand cu. ft. = 3,248,700 cu. ft.

The volume being now determined, from the methods already described it is possible to calculate the Disposable Lift under any conditions whatever.

To summarize, the following are necessary for the ascertainment of Disposable Lift:

Under Conditions of Lift Trial

Available Lift 100 per cent full = Total Weight

= Fixed Weight

Under Conditions in Which Disposable Lift is to be Compared

Available Lift 100 per cent full = Fixed weight

Disposable Lift

Since this memorandum was written, very accurate determinations of the relative weights of hydrogen and air have been made by the British National Physical Laboratory with the result that the lift of 1,000 cu. ft. of hydrogen at 30 in. barometric pressure and at a temperature of 40 deg. Fahr. has been found to be 74.06 lb. consequently the lift formula should be—

72.04 per cent purity a ton is added

Lift per 1,000 cu. ft. =

420 plus Temperature in degrees Fahr.

The foot-candle Lift (Sils) has been shown as ascertained with this determination.

### N. A. C. A. Reports

CONVENTION OF WIND TUNNEL MODELS—Synopsis of Report No. 74, National Advisory Committee for Aeronautics.

This report gives a full description and discussion of the best methods to be employed in constructing models of aircraft and parts thereof for testing in wind tunnels. It includes material on the construction of model assemblies, of fuselages, of airfoil surfaces, and also on the types of models to be used for most special tests as those on pressure distribution. The amount of detail which it is desirable to include on a wind tunnel model is discussed, and the means of approximating the resistance of radiators and other appendages without constructing geometrically similar models are treated. In short, the report is a general handbook on the subject, and is designed to show how to obtain the best results from the often expensive in obtaining models suitable for wind tunnel use.

AIRPLANE ENGINE AND UNITED LIFTING—Synopsis of Report No. 75, National Advisory Committee for Aeronautics.

This report considers the question of possible ranges and useful loads for airplanes from two points of view. A simple calculation is made using a step-by-step method and a theoretical solution is carried through and the two results compared. It is shown that the two methods check each other in a very satisfactory way which is fortunate as the theoretical formulae may be used with confidence and their numerical application is a very simple matter.

It is quite evident that the factors upon which range depends are the lift over drag ratio for the complete machine, the ratio of weight empty to weight fully loaded, the fuel consumption per horsepower hour of the motor and the efficiency of the propeller. The question of an airplane is completely characterized by these factors. The theoretical solution gives explicitly the way these factors contribute to its range and load carrying properties.

The question of range with useful load is considered including a light is an airplane, delivery of load and return without refueling, this problem being of interest with respect to bombing. It is completely solved and a very practical and expeditious method is derived for all possible cases. Among other results are equations giving means to obtain a given distance time required to use up a given weight of fuel at a distance corresponding to a given weight of fuel. The effects of climb at start and glide at end of flight are shown to be powerfully negligible.

Combinations of high speed flight and flight with maximum power are made with greatest range or economical flight, especially with regard to load carrying range and time. A special method for determining the effect of the wind from a wind curve for all wind speeds either with or against the direction of flight is derived and applied and the effect of wind on angle of incidence for most economical flight is determined. Most of the formulae are applicable not only to range flights at most economical angles but also flight at any angle of incidence.

ANALYSIS OF FUEL-AIR MIXTURES—Synopsis of Report No. 76, National Advisory Committee for Aeronautics.

This report, prepared by the staff of the National Advisory Committee for Aeronautics at Langley Field, is concerned chiefly with the assumptions to be made as to the loading on fuselages. A typical "wing and wing" fuselage is analyzed for different sets of assumptions as to loading loads and for three different conditions in flying. In treating drag conditions, all the aerodynamic loads are included and a single stress diagram is shown indicating all the members from nose to tail. The loading loads due to aerodynamic conditions are shown to be very important and to decrease the stresses in the rest of the fuselage by more than 50 per cent in some instances.

The discussion of stress analysis by graphical methods, is, of course, applicable only to "wing-and-wing" fuselages, but the treatment of inertia, damping, and tail loads is equally useful for winged and monocoque designs. As a result of this investigation of loads various recommendations are made as to the distribution of loads in standard testing of fuselages.

Copies of these reports may be obtained upon request from the National Advisory Committee for Aeronautics, Washington, D. C.

### Trade Note

The Aluminum Manufacturing Co., of Fairfield, Conn., announce the completion of their experiments on The Lighter Fuselage Alloy, which has the following physical properties:

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# ANNOUNCEMENT



## Schools

The Curtiss Aeroplane and Motor Corporation has reopened its flying school at Garden City, Long Island, for instruction. Courses in motor and plane construction and repair, and theoretical courses in elementary aerodynamics, cross country flying, instrument reading, etc., will be given in addition to actual flying instruction under the supervision of pilots who have had actual experience in school work.

## Aerial Photography

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## Passenger-Carrying

The Curtiss Aeroplane and Motor Corporation has resumed flying operations at the Curtiss Flying Field at Garden City, Long Island. Dependable Curtiss airplanes, manned by experienced and careful pilots, are available for flights, cross country work, aerial trips over the city, and aerial advertising. Special rates made to parties of five or more people.

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Garden City, Long Island, New York

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The Aerovox Traction Electric Company, Curtiss distributor for Greater New York and Florida, has reopened its school for instruction in free flying at Free Washington, L. I. Flying operators will be under the supervision of David McCullough, pilot of the VAC-1 with Aerovox for

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